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TEACHING TRANSPARENCY WORKSHEET Ionic Bonds

Use with Chapter 7, Section 7.2

1. How many valence electrons does a neutral magnesium (Mg) atom have?
2. What is the charge on a magnesium ion? What does magnesium have to do to form such an ion, and why does it tend to do so?
3. How many valence electrons does a single neutral chlorine (Cl) atom have?
- 4.

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Teaching Transparency Worksheets Chemistry: Matter and Change Chapter 5 . uest 11 Name: Date: Hour: Information: Energy Levels and Sublevels ... Considering your answer to question 5, how many electrons can fit in the entire 5s energy level? can (A_{Vel} is a ha AC5a) 5⁰đ 7. Write the notation for

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an electron spinning clockwise in a p sublevel ...

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TEACHING TRANSPARENCY WORKSHEET The Activity
Series Use with Chapter 9, Section 9.2 1. For each of the
following pairs of elements, underline the one that would
replace the other element in a compound. a. calcium, tin e.
iron, copper b. bromine, fluorine f. iodine, chlorine c.
aluminum, potassium g. silver, lead d. zinc, sodium 2.

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Chapter 19

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Change 0 Chapter 8 . 1.7, the CF bond is polar covalent. (b)
!ENCI"P = ENCI " ENP. = 3.0" 2.1 !ENCI"P = 0.9. Since 0.5 <
EN. A. < 1.7, the PCl bond is polar covalent. (c) !ENCI"Li
Section 8.5 Electronegativity and Polarity 265 Compare and
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TEACHING TRANSPARENCY WORKSHEET Atomic
Orbitals 1. What is the shape of an s orbital? Class 16 Use

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with Chapter 5, Section 5.2 2. What is the relationship between the size of an s orbital and the principal energy level in which it is found? 3. What is the shape of a p orbital? How many p orbitals are there in a ublevel? 4.

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