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TABLES OF CHEMICAL DATA.

Important values,
constants and
standards molar gas
constant $R = 8.31 \text{ J K}^{-1} \text{ mol}^{-1}$ the
Faraday constant F
the Avogadro
constant L the
Planck constant h .
speed of light in a
vacuum c . rest mass
of proton, m_p

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rest mass of neutron, m_n , 1.675×10^{-27} kg

rest mass of electron, m_e , 9.11×10^{-31} kg

elementary charge $e = 1.60 \times 10^{-19}$ C

Faraday constant $F = 96485$ C mol⁻¹

Avogadro constant $N_A = 6.02 \times 10^{23}$ mol⁻¹

Planck constant $h = 6.63 \times 10^{-34}$ J s

speed of light $c = 3.00 \times 10^8$ m s⁻¹

gas constant $R = 8.31$ J K⁻¹ mol⁻¹

universal gas constant $R = 8.31$ J K⁻¹ mol⁻¹

universal gas constant $R = 8.31$ J K⁻¹ mol⁻¹

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