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$\text{CO}_3^{2-}(\text{aq}) + 2 \text{Na}^+(\text{aq}) \rightarrow \text{Na}_2\text{CO}_3(\text{s})$

Concentration of Na^+ ions is: 0.150 mol Na^+
 $\text{CO}_3^{2-} + 2 \times 0.150 \text{ mol Na}^+ = 0.300 \text{ mol Na}^+$

**Answer: Using the stratigraphic section
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