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Textbook problems: Ch. 1: 1.5, 1.7, 1.11, 1.12
1.5 The time-averaged potential of a neutral hydrogen atom is given by $\phi = \frac{q}{4\pi\epsilon_0} \frac{1}{r} (1 + \frac{r}{2a_0})^{-2}$ where q is the magnitude of the electronic charge, and $a_0 = 0.529 \text{ \AA}$, a_0 being the Bohr radius. Find the

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distribution of charge (both continuous and discrete) that will give

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